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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.004 Å R factor = 0.051 wR factor = 0.164 Data-to-parameter ratio = 13.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-(2,6-Dioxo-3,4,5,6-tetrahydro-2*H*pyran-3-yl)-2,3-dihydro-1*H*-isoindole-1,3-dione

The title compound, $C_{13}H_9NO_5$, contains two molecules in the asymmetric unit, in which the 3-substituted dihydropyran-2,6-dione is not planar.

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Comment

The title compound, (I), has attracted attention as an intermediate for the synthesis of glutamine and of γ -dipeptides of glutamic acid (King & Kidd, 1949; Kokai, 1981). More recently, L-theanine was reported as an antitumour agent (Sadzuka *et al.*, 2000, 2002). The title compound has been used as an intermediate for the manufacture of L-theanine (Qian *et al.*, 2005). Here, we report its crystal structure.



In the crystal structure of (I) which has two molecules in the asymmetric unit (Fig. 1), the 3-substituted-dihydro-pyran-2,6dione is not planar, which is as expected. The C-N bond lengths are in the range 1.397 (3)-1.452 (2) Å, which is between the normal value of a C-N single bond and a C=N double bond, due to conjugation effects. All other bond lengths are within normal ranges (Allen *et al.*, 1987).

Experimental

N-Phthaloyl-L-glutamic acid was prepared according to the literature method of Nefkens *et al.* (1960). L-Glutamic acid (14.7 g, 0.1 mol) was reacted with *N*-carboethoxyphthalimide (21.9 g, 0.1 mol) in water (200 ml) with sodium carbonate (23.3 g, 0.21 mol) to give *N*-phthaloyl-L-glutamic acid (yield 19.9 g, 72%). The title compound, (I) (11.9 g) was obtained by reacting *N*-phthaloyl-L-glutamic acid (13.9 g, 0.05 mol) in acetic anhydride (30 ml) under reflux for 20 min. A small quantity (0.1 g) of (I) was dissolved in acetic acid (20 ml) and single crystals suitable for X-ray diffraction were obtained by spontaneous evaporation of the solvent.

Crystal data $C_{13}H_9NO_5$ $M_r = 259.21$ Monoclinic, $P2_1/n$ a = 11.868 (2) Å b = 10.254 (2) Å c = 19.833 (4) Å $\beta = 106.51 (3)^{\circ}$ $V = 2314.1 (8) Å^3$

Z = 8 $D_x = 1.488 \text{ Mg m}^{-3}$ Mo K α radiation $\mu = 0.12 \text{ mm}^{-1}$ T = 293 (2) K Prism, colourless $0.40 \times 0.30 \times 0.30 \text{ mm}$

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Enraf-Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: ψ scan (*SADABS*; Sheldrick, 1996) $T_{\min} = 0.955, T_{\max} = 0.966$ 4544 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.051$ $wR(F^2) = 0.164$ S = 1.074544 reflections 343 parameters 4544 independent reflections 3113 reflections with $I > 2\sigma(I)$ $\theta_{\text{max}} = 26.0^{\circ}$ 3 standard reflections every 200 reflections intensity decay: none

H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.1P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.21 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -0.27 \text{ e } \text{\AA}^{-3}$

All H atoms attached to C atoms were placed in geometric positions and constrained to ride on their parent atoms, with C–H distances in the range 0.93–0.98 Å. They were treated as riding atoms, with $U_{iso}(H) = 1.2U_{eq}(C)$.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.

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References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.



Figure 1

The asymmetric unit of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 35% probability level.

- Enraf–Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf–Nonius, Delft, The Netherlands.
- Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany. King, F. E. & Kidd, D. A. A. (1949). J. Chem. Soc. pp. 3315–3319.
- Kokai, T. K. (1981). Jpn Patent JP 56097260.
- Nefkens, G. H. L., Tesser, G. I. & Nivard, R. J. F. (1960). *Helv. Chim. Acta*, **79**, 688–698.
- Qian, S. S., Chen, R. & Liu, Y. (2005). JingXi HuaGong 22, 845–847. (In Chinese.)
- Sadzuka, Y., Sugiyama, T. & Sonobe, T. (2000). Cancer Lett. 158, 119-124.
- Sadzuka, Y., Yamashita, Y., Kishmoto, S., Fukushima, S., Takeuchi, Y. &
- Sonobe, T. (2002). Pharm. Soc. Jpn, 122, 995-999.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.